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greater electronegativity than atoms that become positive ions. ...

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mythological - solely.

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### **Answer Key, Chemquest 26 ...**

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Hour: \_\_\_\_\_ Information: Definition of  
Electronegativity Electronegativity is a  
measure of how much an atom attracts  
an electron. The higher the  
electronegativity, the greater the atom's  
attraction for electrons. Atoms that  
become negative ions

### **Information : Definition of Electronegativity**

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Therefore, the charge on each X is 1.

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Here's why: Remember Answer: + that S has a negative 2 charge. There are two "X" ions to balance out the negative 2 charge of S.

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